### **Particle Shape**



**Ammonium Sulfate** 

**Potassium Nitrate** 





Sodium Carbonate Monohydrate

Potassium Chloride



**Sulfamic Acid** 

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### **Shape Factor**

- ■A normal way of expressing the shape factor is to make it
	- $\blacksquare$  the ratio of the particle property to the property of a sphere having a diameter equal to the measured particle dimension

#### **Volume-based Shape Factor**

particle volume

volume of a sphere of same diameter

$$
\Psi'_v = \frac{\psi_v D_p^3}{\frac{\pi}{6} D_p^3} = \frac{\psi_v}{\frac{\pi}{6}}
$$

=

### **Sphericity**

#### □ A surface-volume shape factor

#### surface area of the particle surface area of a sphere of volume equal to that of the particle =

## **Sphericity cont…**

 $P$ *PA<sub>P</sub> D A A* 2  $\rm 0$  $\psi = \frac{A_0}{4} = \frac{\pi D_0}{4}$ 

*p p A V* $\binom{p}{2}^{2/3}$  $\pi$ ( $\frac{6}{\pi}$  $\pi$ =

#### Where

D

V p

- $\mathsf{A}_\mathsf{o}$ , $\mathsf{A}_\mathsf{p}$  = surface area of the equivalent sphere and of the particle respectively
	- $=$  diameter of the equivalent sphere
		- = particle volume

### **Ratio Of Specific Surface**

specific surface of the particle

specific surface of a sphere of the same "diameter"

$$
\eta = \frac{\text{specific surface (cm}^2 / \text{gm})}{6D_p}
$$

#### $\Box$  "diameter" is usually taken as the mean screen opening

ρ

=

#### **Ratio Of Specific Surface**



### **Advantage**

**Q** The specific surface of a material for which there are no data may be roughly estimated from the ratio of specific surfaces of a similar material

#### **Specific Surface from Ratio**

Total Surface = 
$$
\frac{6\eta_1 m_1}{\rho(D_p)_1} + \frac{6\eta_2 m_2}{\rho(D_p)_2} + \dots + \frac{6\eta_i m_i}{\rho(D_p)_i} = \frac{6}{\rho} \sum_{i=1}^k \frac{\eta_i m_i}{(D_p)_i}
$$

$$
\text{Average Specific Surface} = \frac{\frac{6}{\rho} \sum_{i=1}^{k} \frac{\eta_i m_i}{(D_p)_i}}{\sum_{i=1}^{k} m_i} = \frac{6}{\rho} \sum_{i=1}^{k} \frac{\eta_i x_i}{(D_p)_i}
$$

### **Bed Porosity**

# $\square$  Fraction void volume  $\mathsf{E}{=}\mathsf{V}_{\mathsf{V}}\mathsf{/}\mathsf{V}_{\mathsf{T}}$  $=$  1-V  $_{\rm P}/$ V  $_{\rm T}$

ρ*AHM P* 1−=

#### **Porosity**

#### $\square$  Porosity of a static bed depends upon

- **Particle shape and surface** roughness
- **Particle size and size distribution**
- Size of the container relative to the particle diameter
- **Method of packing**

### **Method of Packing**

- Water-fill method initially gives more porous packing
- **BUT**

■ Vibration of the vessel and the effect of gas or liquid flow through it ultimately compacts the bed

## **Particle Shape and Surface Roughness**

- $\square$  The lower the particle sphericity, the more open is bed
- □ Particles settle across each other and pack with pointed ends against each other, preventing a close packing

#### **Sphericity as a Function of Porosity**



Figure B-12. Sphericity as a function of porosity for random-packed beds of uniformly sized particles (2). (By permission of John Wiley & Sons, copyright © 1950.)

#### **Particle Size and Size Distribution**

 $\square$  Presence of fine and coarse particles results in a bed of lower porosity than would be obtained with uniform particles

### **Particle Size and Vessel Size**

- $\square$  The presence of container wall interrupts the pattern of particle-toparticle contacts
- $\square$  Hence makes for a larger fraction voids at the wall

#### **Reference**

#### Foust *et al*: **Principles of Unit Operations**, second edition, John Wiley & Sons, Page#711-714

#### **ASSIGNMENT: Problems** B-2, B-3, B-5, B-6, B-7